

Cyclibility dependence on microstructure of ruthenium oxide electrode for thin film micro-supercapacitors

Han-Ki Kim^{*,**,*}, Jae-Hong Lim^{*}, Tae-Yeon Seong^{***,},
 Won Il Cho^{**}, and Young Soo Yoon^{*}

^{*}Thin Film Technology Research Center, ^{**}Battery and Fuel Cell Research Center

Korea Institute of Science and Technology, P. O. Box 131
 Cheongryang, Seoul 130-650, Korea

^{***}Department of Material Science and Engineering,
 Kwangju Institute of Science and Technology,
 Kwangju, 500-712, Korea

In this work, we report cyclibility dependence of RuO₂/LiPON/RuO₂ thin film micro-supercapacitor on microstructure of the RuO₂ electrode film. Both bottom and top RuO₂ films were deposited over Pt current collector with different oxygen gas flow ratio [O₂/(O₂+Ar)] of 10% and 30%, respectively. To investigate the microstructure of the RuO₂ films, which is grown at different oxygen flow ratio, both glancing angle x-ray diffraction (GXR) and transmission electron microscope (TEM) analysis were employed. As shown Fig.1 and Fig.2, the film grown at 10% gas ratio demonstrate amorphous characteristics, while the film grown at 30% gas ratio shows the polycrystalline characteristics. The results of GXR are consistent with selected area diffraction (SAD) pattern shown in Fig.2.

Room temperature charge-discharge measurement based on symmetric RuO₂/LiPON/RuO₂ structure clearly demonstrates the cyclibility dependence on microstructure of RuO₂ electrode in Fig.3. Inset of Fig.3 exhibit plots of voltage vs. time for the TFSC containing the RuO₂ electrodes that were grown at a different oxygen flow ratio of 0.1 and 0.3, respectively. Although the sample grown at oxygen flow 0.1 exhibit higher initial discharge capacity than that grown at 0.3, the TFSC containing RuO₂ electrode grown at a oxygen flow ratio 0.3 show better cycling performance, compared to the sample grown at a flow 0.1. It is thought that the faster degradation of thin film micro-supercapacitor consisting of amorphous RuO₂ electrode is caused by many Li trapping sites, which is related with dangling bond in amorphous structures. In addition, the durability of crystalline RuO₂ film for Li intercalation compared to the amorphous sample plays important role to improve the cyclibility of the thin film supercapacitor having a crystalline RuO₂ electrode.

Using electrochemical and microstructural analysis, we correlate cyclibility of thin film micro supercapacitor with microstructure of RuO₂ electrode.

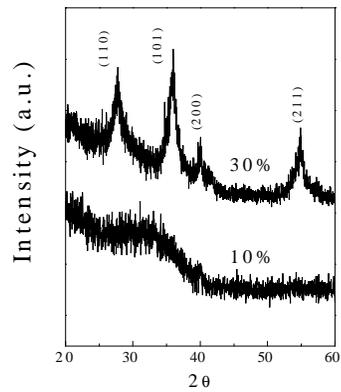


Figure 1. GXR plots of the RuO₂ film grown at different oxygen gas ratio of 10% and 30%, respectively.

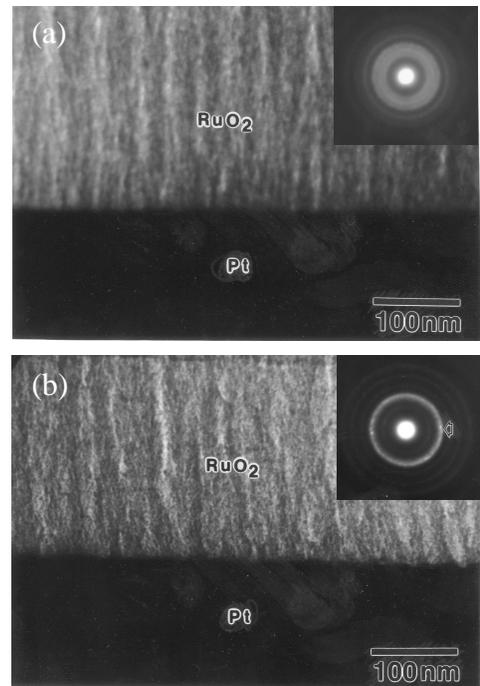


Figure 2. TEM image of the RuO₂ film grown at (a) 10% and 30% oxygen flow ratio with SAD pattern.

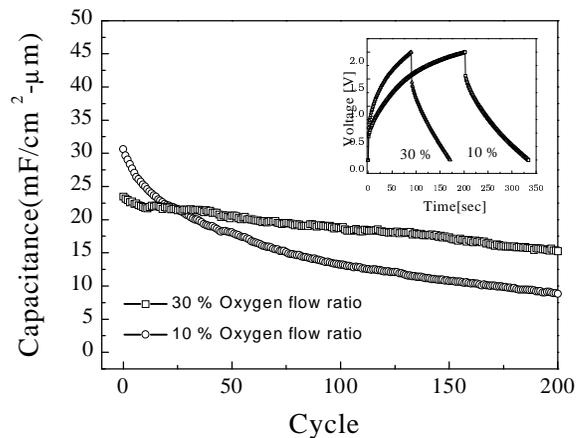


Figure 3. Room temperature charge-discharge profile as a function of cycle number (cycled between 0 and 2 V).